RESULTS

The techniques for determining the melting point and the change in specific volume of melting as a function of pressure have been previously described [23,38]. Figures 1 and 2 demonstrate the effect of pressure on the melting temperature for samples that crystallize in phases I and II, respectively. When the copolymer samples of 93-7 mole% VF₂-TFE and 91-9 mole% VF₂-VF₃ are heated to 225°C and pressure-crystallized at 5000 atm, high-melting planar zig-zag forms result. In a similar manner, if PVF₂ is heated to 285°C and pressure-crystallized at 5500 atm, a high-melting planar zig-zag form results. Although the degree of supercooling necessary to obtain the pure melting planar zig-zag form of the PVF₂ homopolymer is limited, the degree of supercooling requirements for the copolymer samples studied appear to be greatly relaxed. Throughout the pressure range studied, both high-pressure forms of the copolymers



FIG. 1. Effect of pressure on the melting temperature of PVF₂ and copolymers of VF₂ that crystallize in phase I. (1) 93-7 Copolymer VF₂-TFE(LPC); (2) 93-7 copolymer VF₂-TFE(HPC); (3) 95-5 copolymer VF₂-VF; (4) 91-9 copolymer VF₂ VF₃ (HPC); (5) 95-5 blend PVF₂-PVF; (6) PVF₂ from DMSO.



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FIG. 2. Effect of pressure on the melting temperature of PVF_2 and copolymers of VF₂ that crystallize in phase II. (1) 91-9 Copolymer VF_2 - VF_3 (LPC); (2) radiation-polymerized PVF_2 ; (3) Kynar.

as well as the high-pressure-crystallized sample of PVF2 have higher melting points than the corresponding low-pressure-crystallized samples. If these high-melting planar zig-zag forms are melted in the DTA and allowed to recrystallize under atmospheric pressure, they will return to the chain conformation and unit cell dimensions of the low-pressure-crystallized forms. Figures 1 and 2 indicate that within experimental error there exists a linear relationship between the melting temperature and applied pressure (up to 2500 atm) for all the samples examined. This linear behavior has also been previously observed for PVF, phase II and the high-pressure-crystallized mixture of phases I and II [23]. When copolymer samples were melted at pressures below 2500 atm, only a small amount of degradation occurred as evidenced by the samples turning light brown. Since there were no changes in the DTA or in X-ray photographs, it was assumed that the degradation must be minor. However, total decomposition of the sample occurred in numerous

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